



PFAST Team 1 — Occurrence of PFAS in North Carolina's drinking water sources

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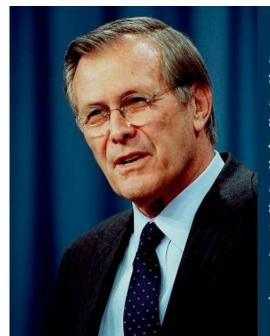
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Key Motivating Question

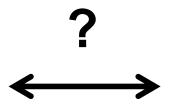
IS MY WATER SAFE TO DRINK?

What do we know?



"There are known knowns. These are things we know that we know. There are known unknowns. That is to say, there are things that we know we don't know. But there are also unknown unknowns. There are things we don't know we don't know we don't know."

Donald Rumsfeld



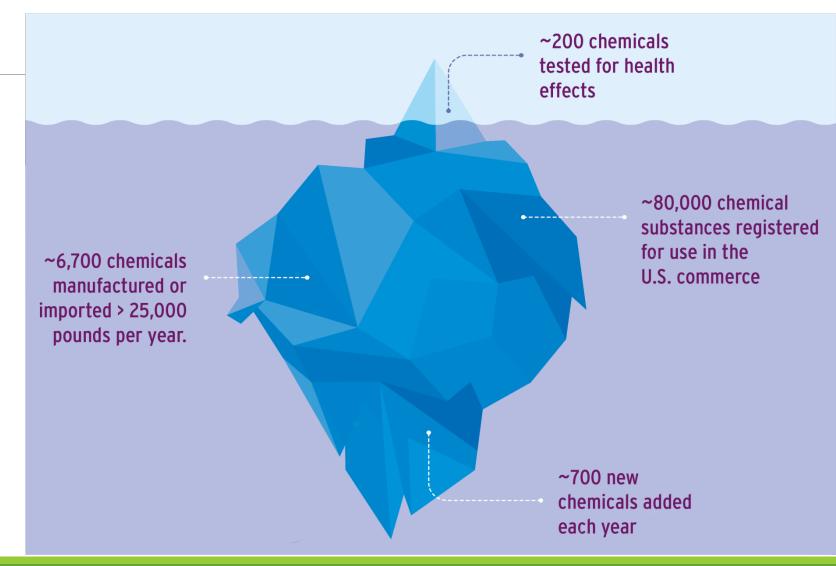


Regulated contaminants (known knowns)

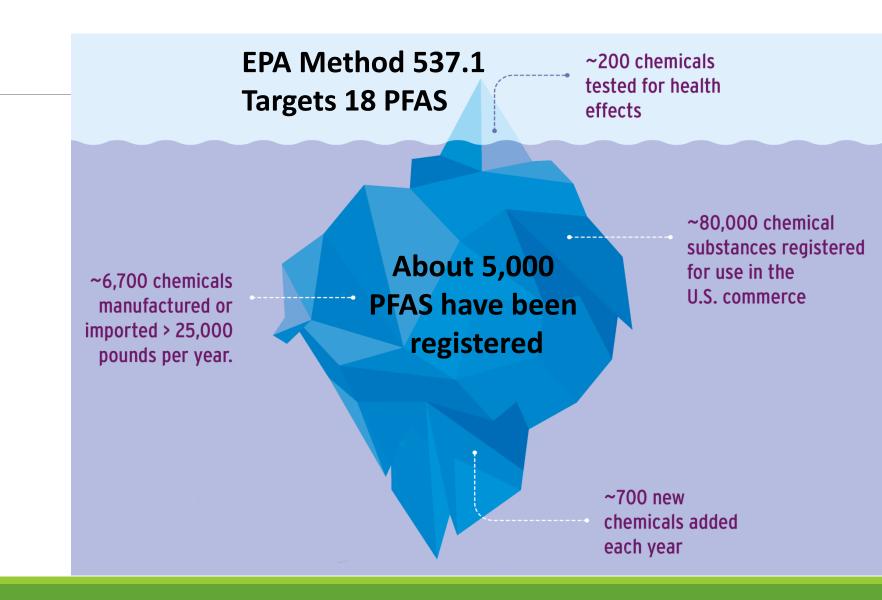
Unregulated contaminants on our radar (known unknowns)

The universe of chemicals (unknown unknowns)

We only know the tip of the iceberg

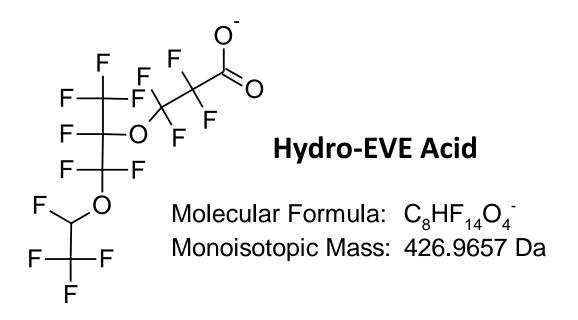


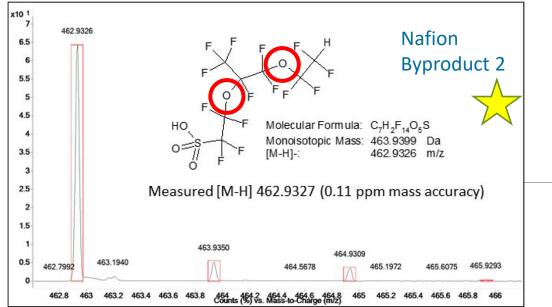
Similar situation with PFAS



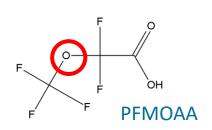
Non-targeted analysis is an emerging approach in environmental analytical chemistry to identify unknown unknowns

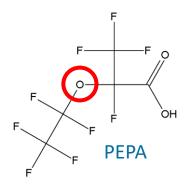


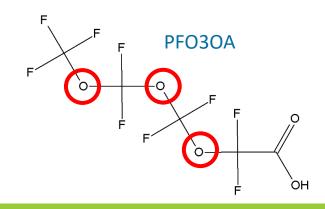


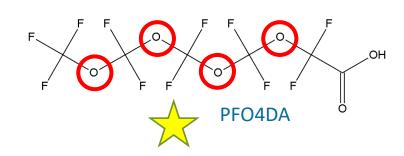


Per- and polyfluoroalkyl ether acids identified by high resolution mass spectrometry











Legislative Mandate: 2018 Appropriations Act (599; SL 2018-5)

FUNDING TO ADDRESS PER- AND POLY-FLUOROALKYL SUBSTANCES, INCLUDING GENX/USE OF EXPERTISE AND TECHNOLOGY AVAILABLE IN INSTITUTIONS OF HIGHER EDUCATION LOCATED WITHIN THE STATE

SECTION 13.1.(f) The General Assembly finds that (i) per- and poly-fluoroalkyl substances (PFAS), including the chemical known as "GenX" (CAS registry number 62037-80-3 or 13252-13-6), are present in multiple watersheds in the State, and impair drinking water and (ii) these contaminants have been discovered largely through academic research not through systematic water quality monitoring programs operated by the Department of Environmental Quality or other State or federal agencies. The General Assembly finds that the profound, extensive, and nationally recognized faculty expertise, technology, and instrumentation existing within the Universities of North Carolina at Chapel Hill and Wilmington, North Carolina State University, North Carolina A&T State University, Duke University, and other public and private institutions of higher education located throughout the State should be maximally utilized to address the occurrence of PFAS, including GenX, in drinking water resources.

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Research Questions

What are the concentrations of targeted legacy and emerging PFAS contaminants in North Carolina public drinking water sources?

- Collect and analyze raw water samples during two consecutive quarters of 2019 at all 191 municipal surface water intakes and all 149 municipal drinking water systems treating groundwater in NC for PFAS measurement
- Repeat this sampling for systems with detectable PFAS in the third quarter of 2019

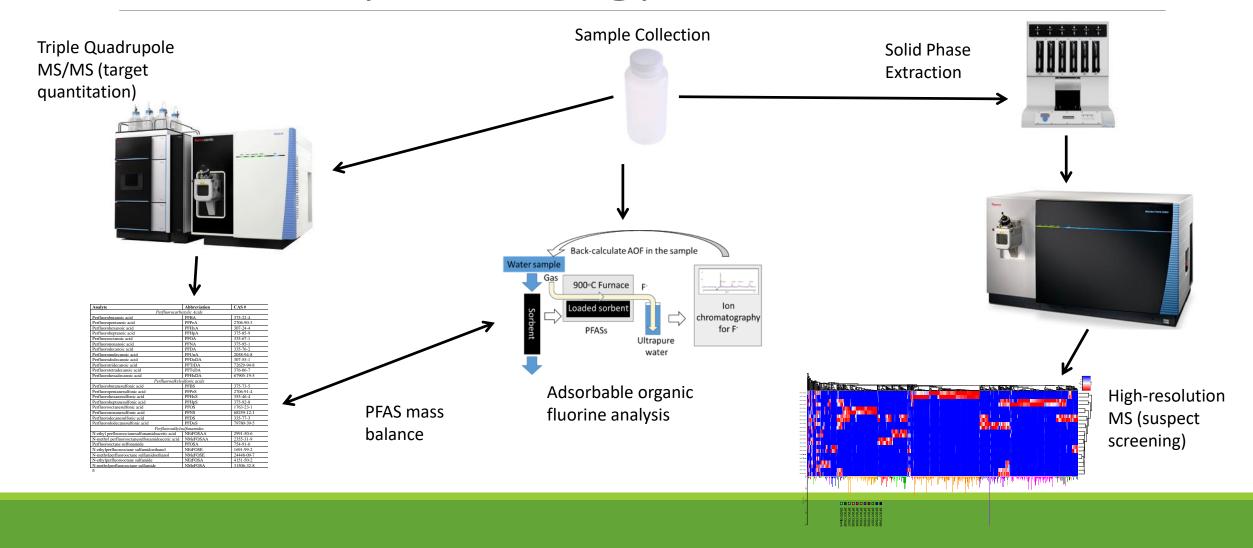
What unanticipated and untargeted PFAS compounds occur in North Carolina public drinking water sources?

Apply high-resolution mass spectrometry methods to screen samples collected above for presence of
 5,100 known PFAS compounds as well as for unknown fluorinated organic compounds

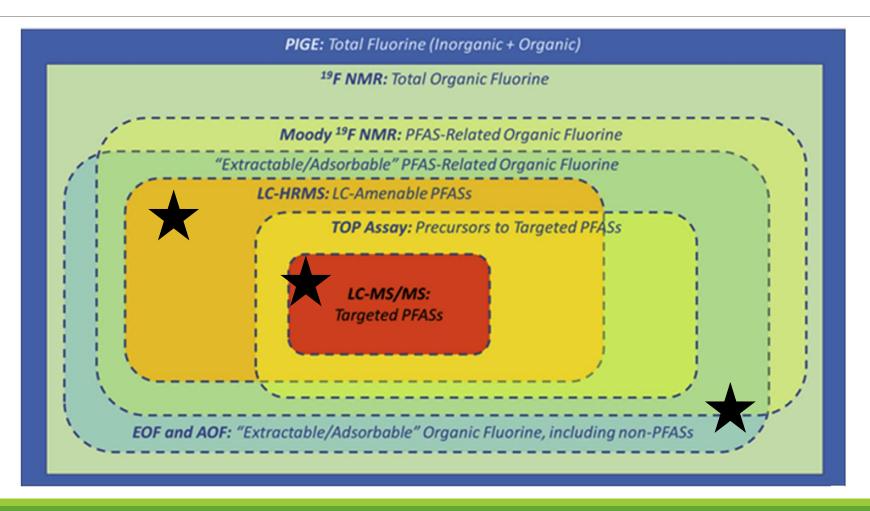
How much of the total organic fluorine in North Carolina public drinking water sources can be accounted for by targeted PFAS quantitation?

 Utilize adsorbable organic fluorine (AOF) measurements in concert with the quantitative PFAS measurements outlined above to assess fluorine "mass balance" in water samples

PFAS analysis strategy



We are using complementary methods to obtain a comprehensive understanding of PFAS occurrence in NC drinking water sources



PFAS target compounds (quantitative measurements) – 55 compounds

Analyte	Abbreviation	CAS#
Perfluorocarb	oxylic Acids	
Perfluorobutanoic acid	PFBA	375-22-4
Perfluoropentanoic acid	PFPeA	2706-90-3
Perfluorohexanoic acid	PFHxA	307-24-4
Perfluoroheptanoic acid	PFHpA	375-85-9
Perfluorooctanoic acid	PFOA	335-67-1
Perfluorononanoic acid	PFNA	375-95-1
Perfluorodecanoic acid	PFDA	335-76-2
Perfluoroundecanoic acid	PFUnA	2058-94-8
Perfluorododecanoic acid	PFDoDA	307-55-1
Perfluorotridecanoic acid	PFTrDA	72629-94-8
Perfluorotetradecanoic acid	PFTeDA	376-06-7
Perfluorohexadecanoic acid	PFHxDA	67905-19-5
Perfluoroalkyls	ulfonic acids	
Perfluorobutanesulfonic acid	PFBS	375-73-5
Perfluoropentanesulfonic acid	PFPeS	2706-91-4
Perfluorohexanesulfonic acid	PFHxS	355-46-4
Perfluoroheptanesulfonic acid	PFHpS	375-92-8
Perfluorooctanesulfonic acid	PFOS	1763-23-1
Perfluorononanesulfonic acid	PFNS	68259-12-1
Perfluorodecanesulfonic acid	PFDS	335-77-3
Perfluorododecanesulfonic acid	PFDoS	79780-39-5
Perfluoroalkyls	sulfonamides	
N-ethyl perfluorooctanesulfonamidoacetic acid	NEtFOSAA	2991-50-6
N-methyl perfluorooctanesulfonamidoacetic acid	NMeFOSAA	2355-31-9
Perfluorooctane sulfonamide	PFOSA	754-91-6
N-ethylperfluorooctane sulfamidoethanol	NEtFOSE	1691-99-2
N-methylperfluorooctane sulfamidoethanol	NMeFOSE	24448-09-7
N-ethylperfluorooctane sulfamide	NEtFOSA	4151-50-2
N-methylperfluorooctane sulfamide	NMeFOSA	31506-32-8

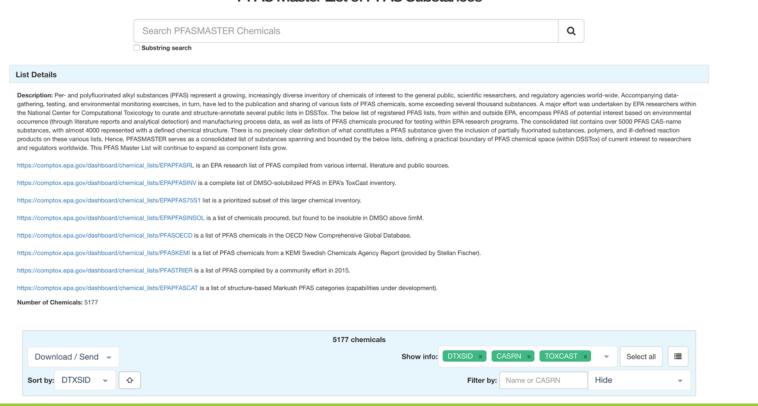
Analyte	Abbreviation	CAS#
Fluorotelomer s	ulfonic acids	
4:2 Fluorotelomer sulfonic acid	4:2 PFS	757124-22-4
6:2 Fluorotelomer sulfonic acid	6:2 PFS	27619-97-2
8:2 Fluorotelomer sulfonic acid	8:2 PFS	39108-34-4
10:2 Fluorotelomer sulfonic acid	10:2 PFS	120226-60-0
Perfluoroalkyl ether carbo.	xylic and sulfonic acids	
Perfluoro-2-propoxypropanoic acid	GenX	13252-13-6
Dodecafluoro-3H-4,8-dioxanonanoic acid	ADONA	958445-44-8
9-chlorohexadecafluoro-3-oxanonane-1-sulfonate	F-53B (Major)	73606-19-6
11-chloroeicosafluoro-3-oxanonane-1-sulfonate	F-53B (Minor)	83329-89-9
Perfluoro-2-methoxyacetic acid	PFMOAA	674-13-5
Perfluoro-2-methoxypropanoic acid	PMPA	13140-29-9
Perfluoro-2-ethoxypropanoic acid	PEPA	N/A
Perfluoro(3,5-dioxahexanoic) acid	PFO2HxA	39492-88-1
Perfluoro(3,5,7-trioxaoctanoic) acid	PFO3OA	39492-89-2
Perfluoro(3,5,7,9-tetraoxadecanoic) acid	PFO4DA	39492-90-5
Perfluoro(3,5,7,9,11-pentaoxadodecanoic) acid	PFO5DoDA	39492-91-6
Ethanesulfonic acid, 2-[1-[difluoro[(1,2,2-	Nafion by-product 1	29311-67-9
trifluoroethenyl)oxy]methyl]-1,2,2,2-		
tetrafluoroethoxy]-1,1,2,2-tetrafluoro-		
Ethanesulfonic acid, 2-[1-[difluoro(1,2,2,2-	Nafion by-product 2	749836-20-2
tetrafluoroethoxy)methyl]-1,2,2,2-		
tetrafluoroethoxy]-1,1,2,2-tetrafluoro-		
2,2,3,3,4,5,5,5-4-(1,1,2,2-tetrafluoro-2-	Nafion by-product 4	N/A
sulfoethoxy)pentanoic acid		
Propanoic acid, 3-[1-[difluoro(1,2,2,2-	Hydro-EVE acid	773804-62-9
tetrafluoroethoxy)methyl-1,2,2,2-		
tetrafluoroethoxy]-2,2,3,3-tetrafluoro-		
1,1,2,2-tetrafluoro-2-(1,2,2,2-tetrafluoro-	NVHOS	N/A
ethoxy)ethane sulfonic acid		

"Semi-Targeted" high-resolution MS screening for PFAS



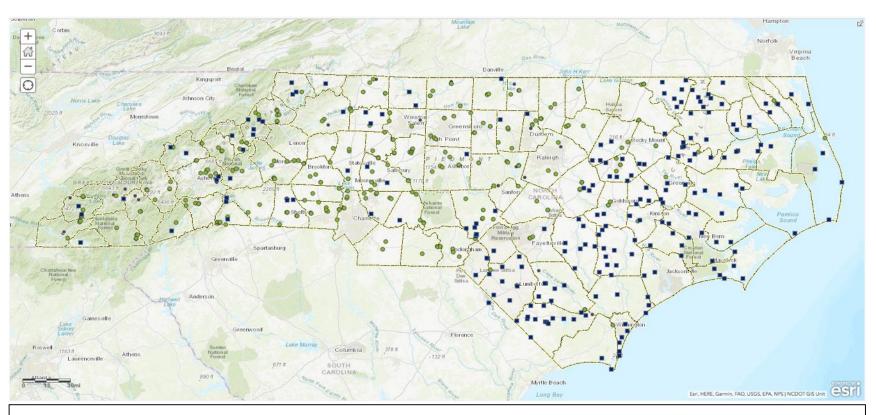
PFAS Master List of PFAS Substances

- "Master List" of > 5,000
 PFAS compounds
- Data will be nonquantitative (relative abundances among water samples)
- Potential to discover novel PFAS contaminants



Sample acquisition strategy

- 191 Municipal surface water sources
- 149 Municipal well water sources



Surface (green circle) and groundwater (blue square) sampling sites for drinking water sources to be analyzed for PFAS compounds.

PFAS analysis strategy

Sample collection:

- Coordinated between NCSU and Duke
- ~25-30 samples per week, organized in geographic sectors
- Samples collected in polypropylene bottles and stored on ice during transport

QA/QC and replication:

- Trip blanks included with every sampling
- Trip spikes (50 ng/L analyte addition) included with every sampling
- Duplicate samples for 10% of sites
- NCSU & Duke will split sample analyses evenly, with 20% samples analyzed by both labs

Team 1

Duke University

Lee Ferguson, PhD Co-Lead

Abigail Joyce, PhD

- Sampling
- Quantitative Analysis

Gordon Getzinger, PhD

Non-target Analysis

NC State University

Detlef Knappe, PhD Co-Lead

Noelle DeStefano, PhD

- Sampling
- Quantitative Analysis
- Non-Target Analysis

Zachary Hopkins

Quantitative Analysis

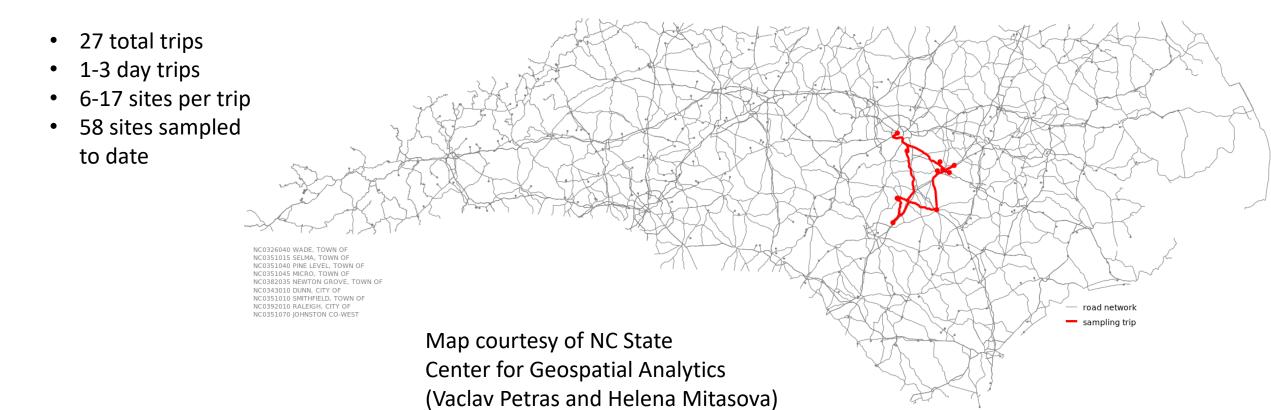
UNC - Charlotte

Mei Sun, PhD Co-Investigator

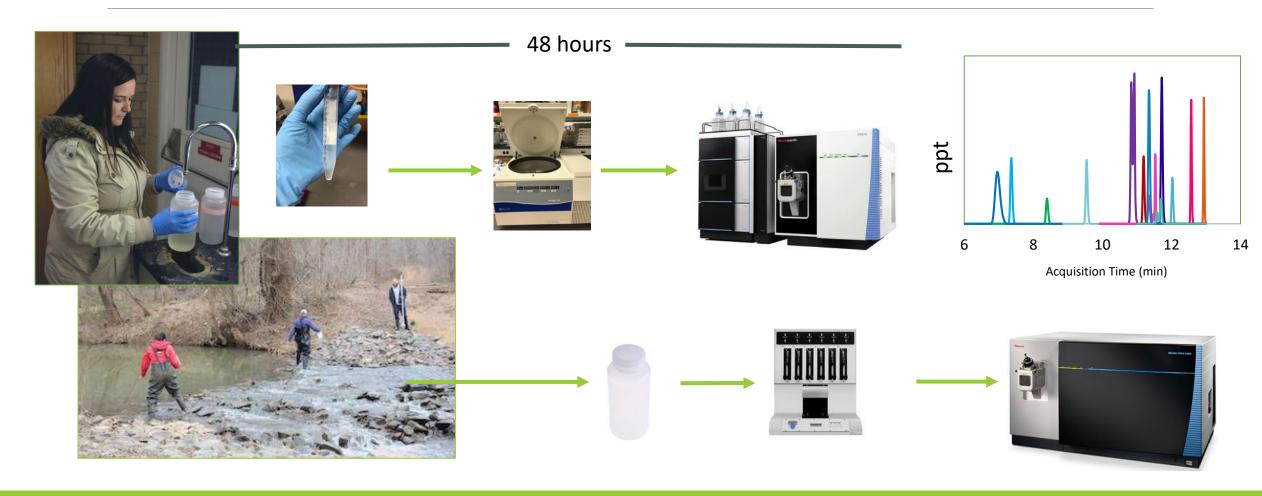
Yuling Han, PhD Vivek Pulikkal

- Adsorbable organic fluorine (AOF) measurements
- Extractable organic fluorine (EOF) measurements

Sampling trip design



Sample Processing – LC/MS



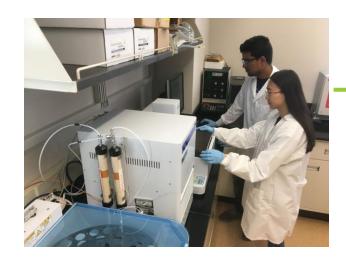
Sample Processing – AOF/EOF



Organic fluorine adsorption by activated carbon (AC)



Organic fluorine adsorption by solid phase extraction (SPE)

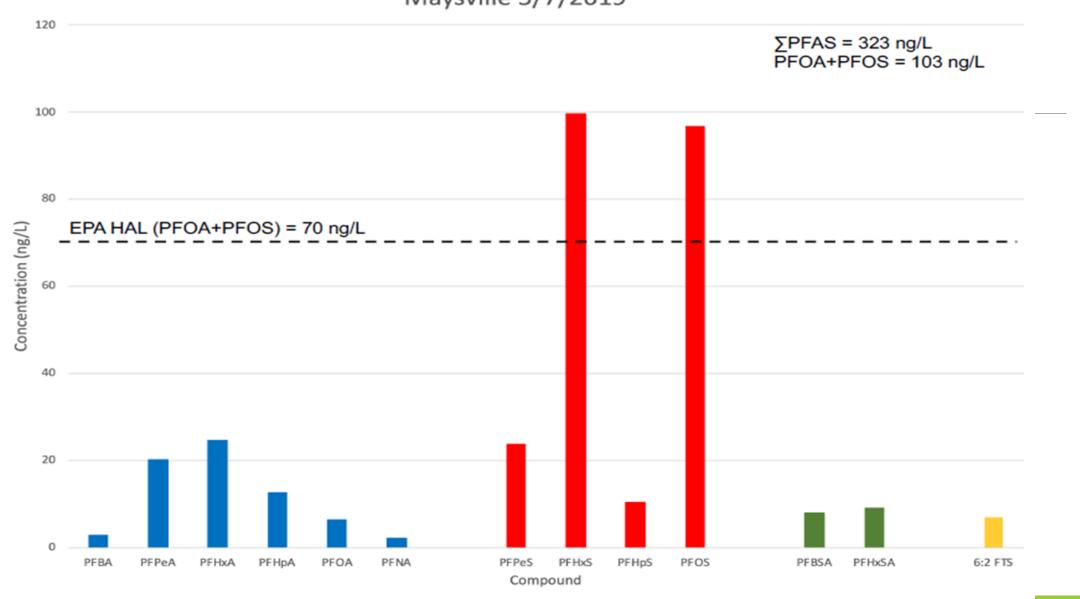


Combustion of AC or SPE concentrate



Fluoride analysis by ion chromatography





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- PFAST Management Team
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